

Stoichiometry Test Holt

Stoichiometry Test Holt - Holt Chemistry Chapter 9: Stoichiometry Chapter Exam Instructions. Choose your answers to the questions and click 'Next' to see the next set of questions. Holt McDougal Modern Chemistry Chapter 9: Stoichiometry / Practice Exam Exam Instructions: Choose your answers to the questions and click 'Next' to see the next set of questions. Stoichiometry Chemistry Quiz. This is 1 mole of water produced per mole of P_2O_5 . There are 4 moles of water produced per mole of P_2O_5 . There are 6 moles of water produced per mole of P_2O_5 . There are 10 moles of water produced per mole of P_2O_5 . A compound has an empirical formula is NP_2Cl_5 and a molecular weight of 347.66. Remember it is a MC test, use the answers ... Practice Test Ch 3 Stoichiometry Name _____ Per _____ $2MnO_2 + 4KOH + O_2 + Cl_2 \rightarrow 2KMnO_4 + 2KCl + 2H_2O$. For the reaction above, there is 100. g of each reactant available. Which reagent is the limiting reagent? [Molar Masses: MnO